Part of Paper No. 20050404 for 09/942,008

copy of claims from copending application 09/961, 895 dated
Please amend the claims so that they read as follows:

8/13/2004

1. (Currently Amended) A method for the preparation of a cathode active material comprising a mixing step of mixing starting materials for synthesis of a compound having the formula  $Li_x$ FePO<sub>4</sub>, where  $0 < x \le 1$ ,

a milling step of simultaneously pulverizing and mixing a mixture resulting from said mixing step; and

a sintering step of firing the mixture resulting from said milling step; wherein

a carbon material is added at any one of the above steps;

 $\text{Li}_3\text{PO}_4$  and  $\text{Fe}_3(\text{PO}_4)_2$  or a hydrate thereof  $\text{Fe}_3(\text{PO}_4)_2 \cdot \text{nH}_2\text{O}$ , where n denotes the number of hydrates, are used as said starting materials for synthesis; and wherein

the content of  $Fe^{3+}$  in the total iron in said  $Fe_3(PO_4)_2$  or a hydrate thereof  $Fe_3(PO_4)_2$  nH<sub>2</sub>O, where n denotes the number of hydrates, is set to 61 wt% or less and not less than 2 wt%.

- 2. (Original) The method for the preparation of a cathode active material according to claim 1 wherein the carbon content per unit volume of a  $Li_x$ FePO4 carbon composite material composed of said  $Li_x$ FePO<sub>4</sub>, where  $0 < x \le 1$ , and said carbon material, is not less than 3 wt%.
- 3. (Original) The method for the preparation of a cathode active material according to claim 2 wherein, in the carbon material forming said Li<sub>x</sub>FePO<sub>4</sub> carbon composite material, the strength area ratio A (D/G) of diffraction rays appearing at the number of waves of 1570 to 1590 cm<sup>-1</sup> with respect to diffraction lines appearing at the number of waves of the Raman spectrum of graphite in the Raman spectrographic method is not less than 0.3.
- 4. (Previously Amended) The method for the preparation of a cathode active material according to claim 2 wherein the powder density of said Li<sub>x</sub>FePO<sub>4</sub> carbon composite material is not less than 2.2 gm/cm<sup>3</sup>.

- 5. (Previously Amended) The method for the preparation of a cathode active material according to claim 2 wherein the Bulnauer Brunauer Emmet Teller specific surface area of said Li<sub>x</sub>FePO<sub>4</sub> carbon composite material is not less than 10.3 m<sup>2</sup>/g.
- 6. (Original) The method for the preparation of a cathode material according to claim 2 wherein the first-order particle of said  $Li_x$ FePO<sub>4</sub> carbon composite material is not larger than 3.1  $\mu$ m.
- 7. (Currently Amended) A method for the preparation of a non-aqueous electrolyte cell having a cathode having a cathode active material, an anode having an anode active material and a non-aqueous electrolyte, said method comprising a mixing step of mixing starting materials for synthesis of a compound having the formula  $Li_xFePO_4$ , where  $0 < x \le 1$ .

a milling step of simultaneously pulverizing and mixing a mixture resulting from said mixing step; and

a sintering step of firing the mixture resulting from said milling step; wherein

a carbon material is added at any one of the above steps;

 $\text{Li}_3\text{PO}_4$  and  $\text{Fe}_3(\text{PO}_4)_2$  or a hydrate thereof  $\text{Fe}_3(\text{PO}_4)_2 \cdot \text{nH}_2\text{O}$ , where n denotes the number of hydrates, are used as said starting materials for synthesis; and wherein

the content of Fe<sup>3+</sup> in the total iron in said Fe<sub>3</sub>(PO<sub>4</sub>)<sub>2</sub> or a hydrate thereof Fe<sub>3</sub>(PO<sub>4</sub>)<sub>2</sub>·  $nH_2O$ , where n denotes the number of hydrates, is set to 61 wt% or less and not less than 2 wt%.

8. (Original) The method for the preparation of a non-aqueous electrolyte cell according to claim 7 wherein the carbon content per unit volume of a  $Li_xFePO_4$  carbon composite material composed of said  $Li_xFePO_4$ , where  $0 < x \le 1$ , and said carbon material, is not less than 3 wt%.

- 9. (Original) The method for the preparation of a non-aqueous electrolyte cell according to claim 8 wherein, in the carbon material forming said Li<sub>x</sub>FePO<sub>4</sub> carbon composite material, the strength area ration A (D/G) of diffraction rays appearing at the number of waves of 1570 to 1590 cm<sup>-1</sup> with respect to diffraction lines appearing at the number of waves of 1340 to 1360 cm<sup>-1</sup> of the Raman spectrum of graphite in the Raman spectrographic method is not less than 0.3.
- 10. (Previously Amended) The method for the preparation of a non-aqueous electrolyte cell according to claim 2 wherein the powder density of said Li<sub>x</sub>FePO<sub>4</sub> carbon composite material is not less than 2.2 gm/cm<sup>3</sup>.
- 11. (Original) The method for the preparation of a non-aqueous electrolyte cell according to claim 8 wherein the Bulnauer Emmet Teller specific surface area of said Li<sub>x</sub>FePO<sub>4</sub> carbon composite material is not less than 10.3 m<sup>2</sup>/g.
- 12. (Original) The method for the preparation of a non-aqueous electrolyte cell according to claim 8 wherein the first-order particle of said  $Li_x$ FePO<sub>4</sub> carbon composite material is not larger than 3.1  $\mu$ m.
- 13. (Original) The method for the preparation of a non-aqueous electrolyte cell according to claim 8 wherein said non-aqueous electrolyte is a liquid-based electrolyte employing a non-aqueous electrolyte solution composed of an electrolyte dissolved in a non-aqueous solvent.
- 14. (Original) The method for the preparation of a non-aqueous electrolyte cell according to claim 8 wherein said non-aqueous electrolyte is a solid electrolyte.
- 15. (Original) The method for the preparation of a non-aqueous electrolyte cell according to claim 14 wherein said solid electrolyte is composed of an electrolyte salt and a high molecular compound dissolving said electrolyte salt and wherein said high molecular compound is a gelated electrolyte matrix gelated on absorbing said non-aqueous electrolyte solution.

16. (Currently Amended) A producing method of  $Li_xFePO_4$ , (0 < x ≤ 1) comprising the steps of:

mixing FeSO<sub>4</sub> and phosphate into Fe<sub>3</sub>(PO<sub>4</sub>)<sub>2</sub> wherein the content of Fe<sup>3+</sup> in the total iron in said Fe<sub>3</sub>(PO<sub>4</sub>)<sub>2</sub> is not more than 61 wt% and not less than 2 wt%; mixing said Fe<sub>3</sub>(PO<sub>4</sub>)<sub>2</sub> and Li<sub>3</sub>PO<sub>4</sub> into a mixture; adding carbon material to the mixture; sintering the mixture; and milling the mixture.